

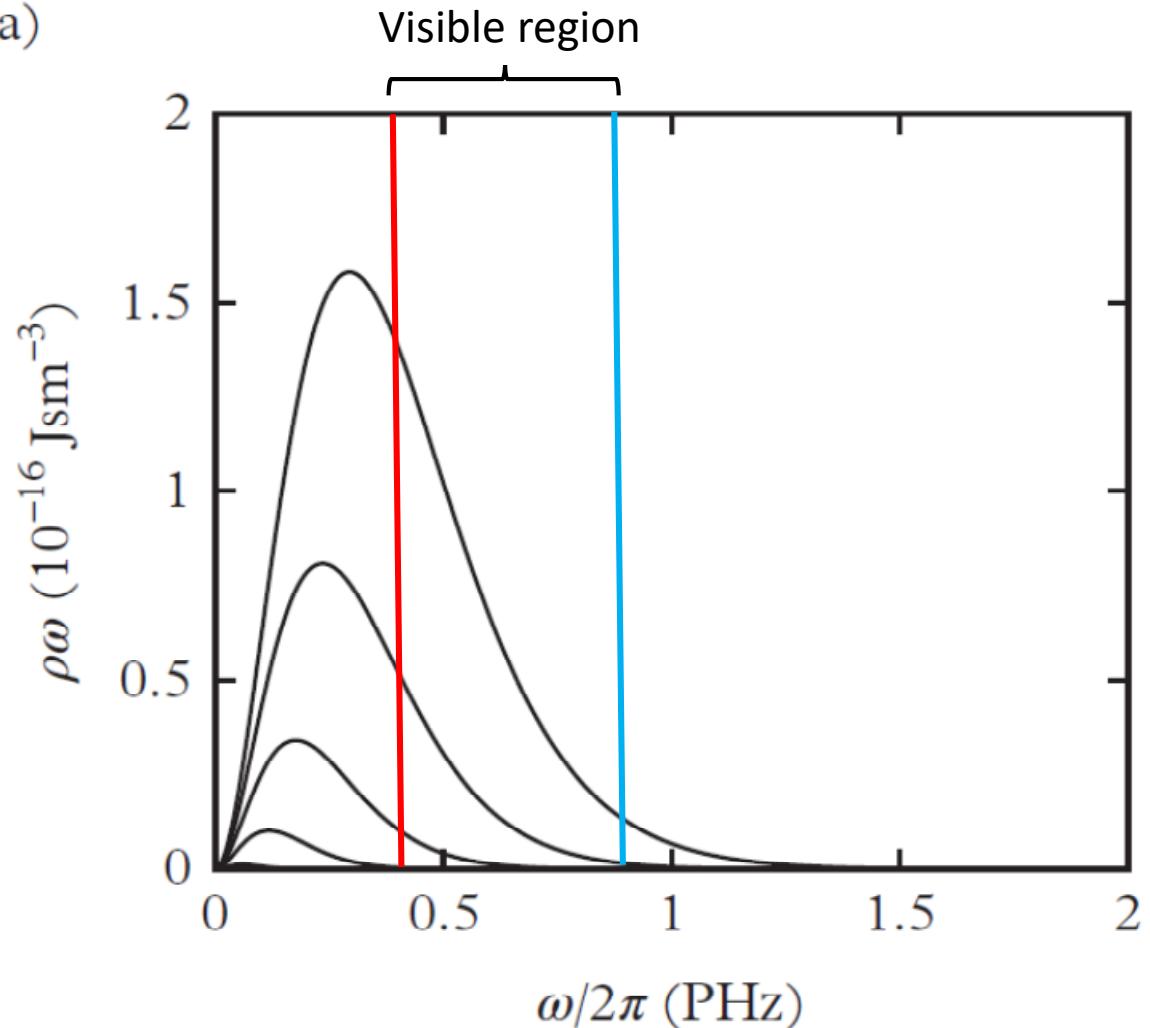
## Thermodynamics lecture 9. Thermal radiation in more detail

1. Wien's argument to show adiabatic expansion of cavity radiation preserves its thermal character
2. Wien's laws:
  1. Wien's displacement law
  2. Wien's distribution law
3. Statistical mechanics of cavity radiation,  
Model 1: distinguishable modes (plane waves)
4. Energy, partition function, etc.
5. Model 2: indistinguishable photons  
(conceptually harder but equally important)

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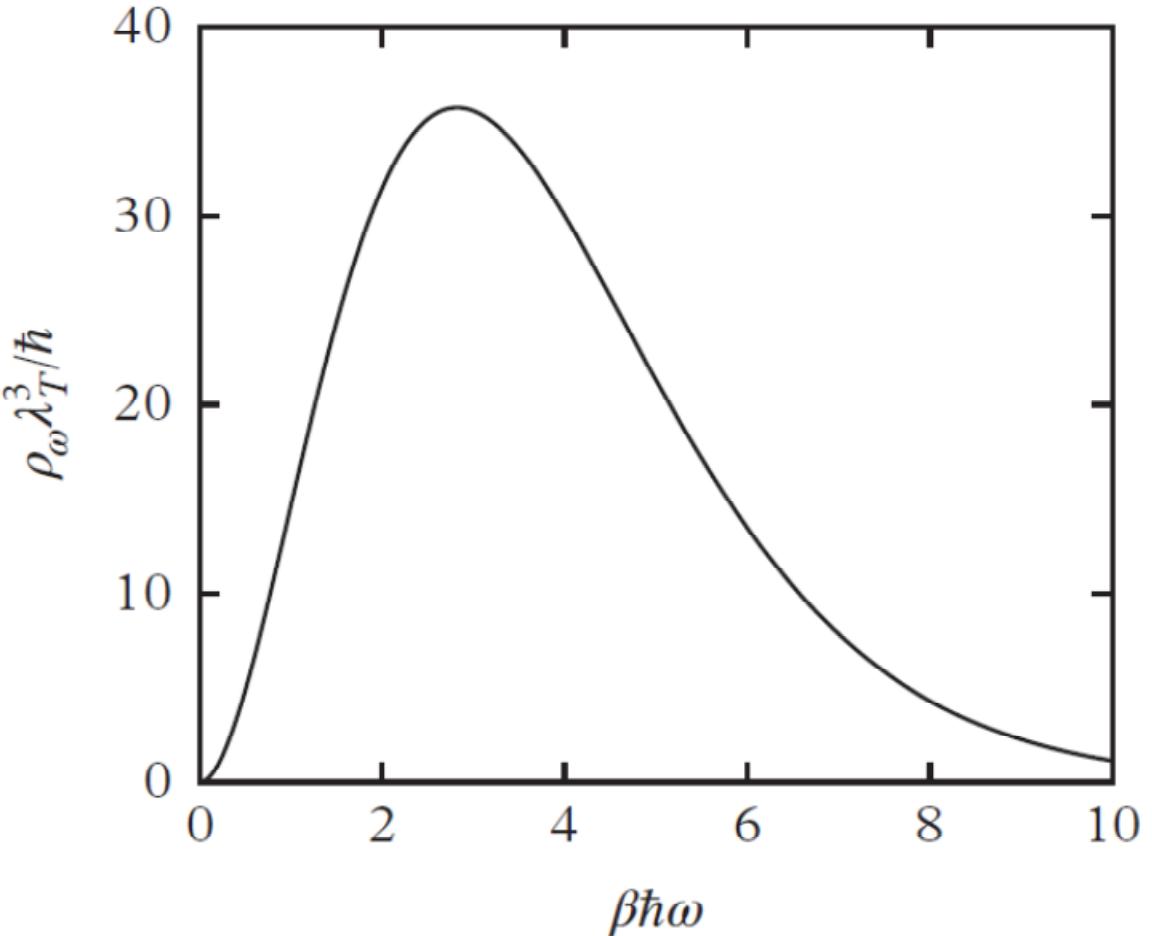
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(a)



Spectral energy density at  
1000, 2000, 3000, 4000, 5000 K

(b)



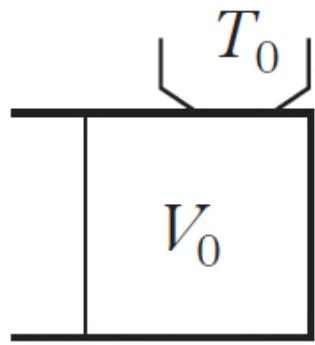
It's the same function each time,  
scaled by  $\lambda_T^3$

$$\lambda_T = 2\pi \hbar c / k_B T.$$

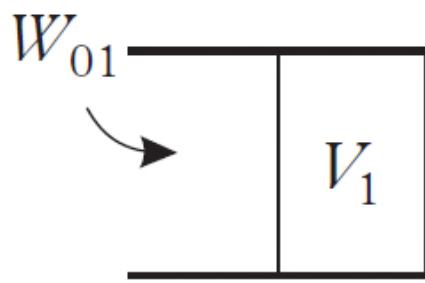
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Slow expansion of cavity radiation in a reflecting cavity:  
does it remain in thermal equilibrium state?

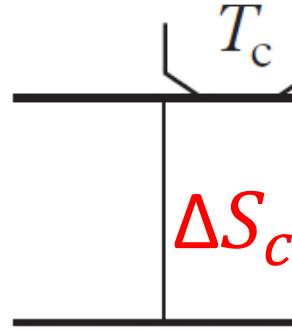
Wien's argument to show the answer is yes:



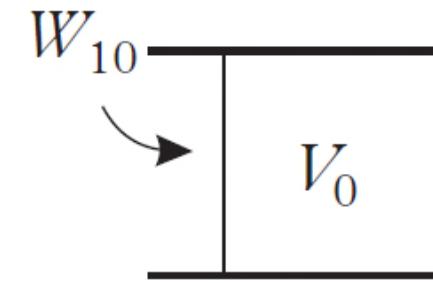
(a)



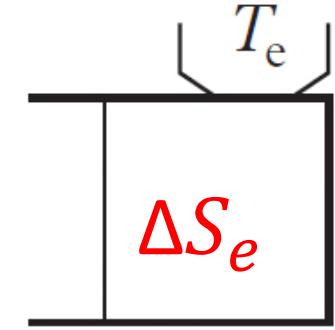
(b)



(c)



(d)



(e)

Stages (c) and (e): we pick  $T_c$  and  $T_e$  such that the radiation comes to equilibrium with **no net  $\Delta U$**

Stages (b) and (d):

Adiabatic change of volume (so work is done and **U changes and changes back**)

**Argument:**

$$\Delta U = 0 \rightarrow \Delta S_c \geq 0, \Delta S_e \geq 0$$

But no net change in state  $\rightarrow$

$$\Delta S_{tot} = 0$$

$$\rightarrow \Delta S_c = \Delta S_e = 0$$

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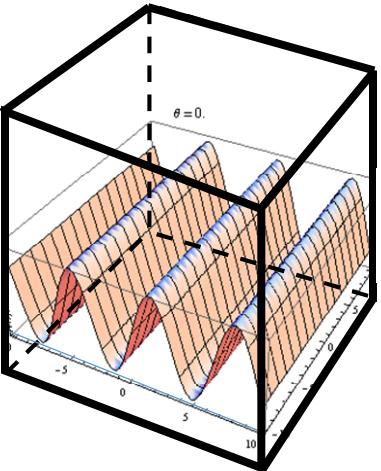
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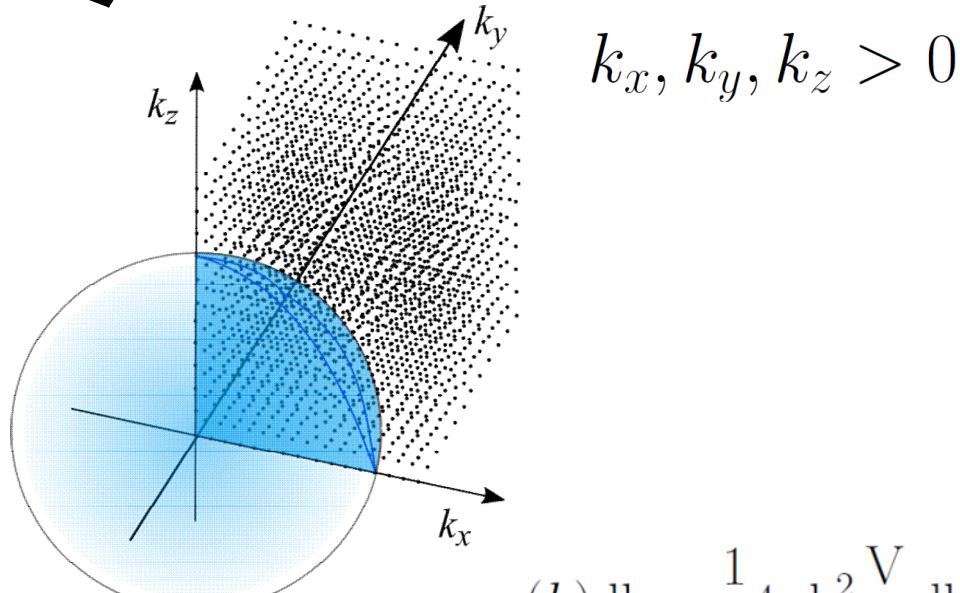
# Modes of electromagnetic field

## Standing waves in a box



$$\sin(k_x x) \sin(k_y y) \sin(k_z z)$$

$$\Delta k_x = \frac{\pi}{L}$$

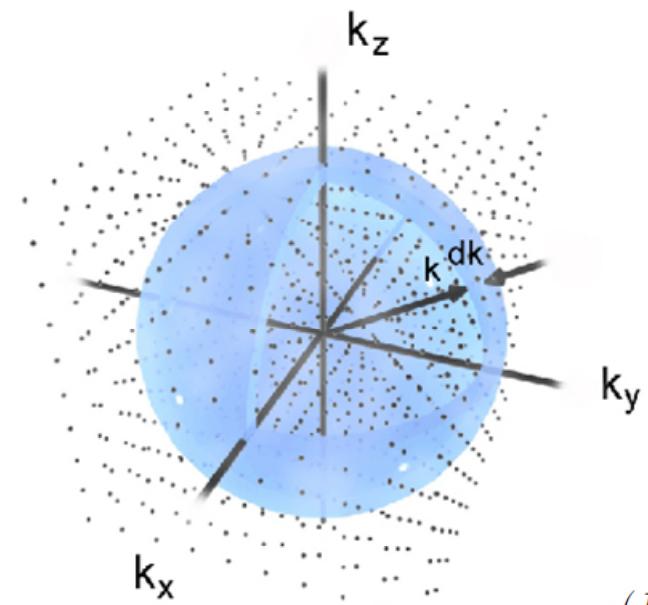


$$g(k)dk = \frac{1}{8}4\pi k^2 \frac{V}{\pi^3} dk$$

OR travelling waves with a *mathematical constraint*: must have period  $L$ .

Travelling waves,  
 $e^{ik_x x} e^{ik_y y} e^{ik_z z}$

$$\Delta k_x = \frac{2\pi}{L}$$



$$g(k)dk = 4\pi k^2 \frac{V}{(2\pi)^3} dk$$

# Comparison between cavity radiation and an ideal gas

	Ideal gas	Cavity radiation
Independent variables	$U, N, V$	$U, V$
Number of particles	$N$	$N = bVT^3$
Equation of state	$p = nk_B T$	$p = 0.9 nk_B T$
Energy	$U = \frac{1}{\gamma-1} N k_B T,$	$U = 2.7 N k_B T,$
$u$ and $p$	$p = (\gamma - 1)u$	$p = \frac{1}{3}u$
Entropy	$S = N k_B \left( \ln \left[ a \frac{T^{1/(\gamma-1)}}{n} \right] \right)$ $= \frac{U}{T} (\ln T - (\gamma - 1) \ln(n/a))$	<div style="border: 2px solid red; padding: 2px;"><math>S = 3.6 N k_B</math></div> $= \frac{4}{3} \frac{U}{T}$
Chemical potential	$\mu = (U - TS + pV)/N$ $= (\gamma u - Ts)/n$	$\mu = 0$

$$b \simeq 2.03 \times 10^7 \text{ m}^{-3} \text{K}^{-3}$$